

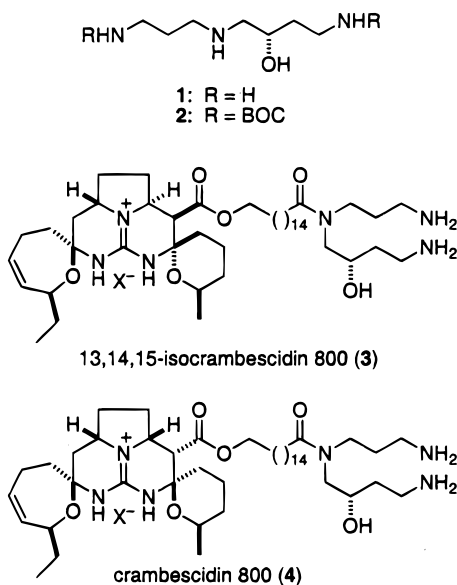
Preparation of (3*S*)-*N,N*-1,7-Bis-*tert*-butoxycarbonyl-3-hydroxyspermidine in High Enantiomeric Purity

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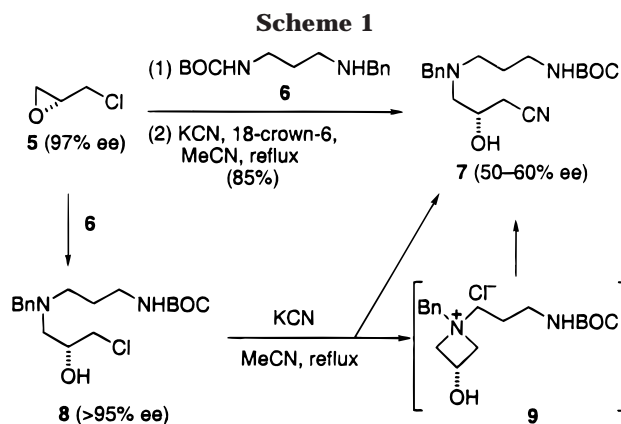
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Although polyamines are structural features of many natural products,¹ 3-hydroxyspermidine (**1**) is rare, being found only in the crambescidin family of guanidinium alkaloids.² During our recent total syntheses of 13,14-, 15-isocrambescidin **800** (**3**)³ and crambescidin **800** (**4**),⁴ we required a practical way to prepare suitably protected 3-hydroxyspermidines in high enantiopurity. Ponasik and Ganem had previously reported the only synthesis of the 3-hydroxyspermidine unit, albeit in racemic form.⁵ Herein we describe a convenient, high-yielding procedure for preparing (3*S*)-*N,N*-1,7-bis-*tert*-butoxycarbonyl-3-hydroxyspermidine (**2**).



(*R*)-Epichlorohydrin (**5**) is an obvious starting material for preparing the (*S*)-1,4-diamino-2-butanol fragment of **2** (Scheme 1), since Bergeron and co-workers had previ-



ously used this precursor to fashion the diaminobutanol unit of (+)-hypusine.⁶ Reaction of [*N*-(*tert*-butoxycarbonyl)-3-aminopropyl]benzylamine (**6**)⁷ and **5** (97% ee) at room temperature for 24 h and then heating the resulting crude chlorohydrin at reflux in acetonitrile in the presence of KCN and 18-crown-6^{6a} provided β -hydroxy nitrile **7** in 85% yield. However, ¹⁹F NMR analysis of the (*R*)- α -methoxy- α -(trifluoromethyl)phenylacetate [(*R*)-MTPA]⁸ derivative revealed that **7** had been generated in only 50–60% ee. To confirm that racemization had occurred in the second step, chlorohydrin **8** was isolated and shown to have high enantiopurity (>95% ee).⁸ Thus, competitive formation of achiral azetidinium salt **9** in refluxing acetonitrile during the second step was undoubtedly the cause of the partial epimerization observed in forming **7**.^{6c,9,10}

In their synthesis of hypusine, Bergeron and co-workers solved the racemization problem by preparing the *N*-benzyloxycarbonyl derivative of the chlorohydrin intermediate prior to reaction with cyanide.^{6c} An alternate solution would be to avoid formation of azetidinium salt **9** during the conversion of **8** \rightarrow **7**. We reasoned that simply switching to a nonpolar solvent for the cyanide displacement step would accomplish this goal. To this end, a solution of [*N*-(*tert*-butoxycarbonyl)-3-aminopropyl]benzylamine (**6**) and (*R*)-epichlorohydrin (**5**) was maintained at room temperature for 24 h, and the resulting crude chlorohydrin was heated at 70 °C in toluene with a slight excess of tetrabutylammonium cyanide for 5 h to give **7** in 80% yield (Scheme 2). Gratifyingly, the enantiomeric purity of **7** formed in this way was ~97% ee.⁸ Selective reduction of the nitrile group of **7** with

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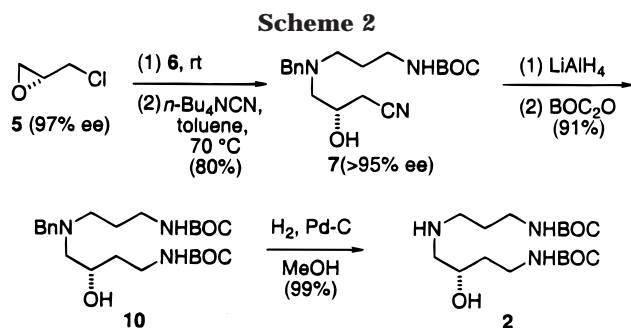
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(9) The intermediacy of azetidinium salts in reactions of 1-chloro-3-dialkylamino-2-propanol derivatives with nucleophiles, including cyanide, is well-known.¹⁰

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lithium aluminum hydride at 0 °C, followed by treatment of the resulting primary amine with di-*tert*-butyl dicarbonate (BOC₂O), provided **10** in 91% yield. Standard hydrogenolysis of the benzyl group of **10** then delivered **2** in nearly quantitative yield.

In summary, a convenient procedure for preparing bis-BOC derivative **2** of (*S*)-3-hydroxyspermidine in high enantiopurity from commercially available (*R*)-epichlorohydrin (**5**) has been developed. This first enantiospecific synthesis of the (*S*)-3-hydroxyspermidine unit proceeds in 72% yield from **5** by way of two isolated and purified intermediates. Partial racemization in generating β-hydroxy nitrile intermediate **7** was prevented by carrying out the displacement reaction of 1-chloro-3-dialkylamino-2-propanol **8** with cyanide in toluene. Since epichlorohydrin is a common precursor of diverse 1-dialkylamino-2-propanol derivatives, performing the nucleophilic displacement of 1-chloro-3-dialkylamino-2-propanol intermediates in a nonpolar solvent should be a useful tactic for enantiospecific synthesis of other epichlorohydrin-derived amino alcohols.

Experimental Section

General Methods. (*R*)- and (*S*)-epichlorohydrin (97% ee) were purchased from Aldrich Chemical Co. Dry THF, Et₂O, and CH₂-Cl₂ were filtered through a column charged with Al₂O₃.¹¹ Triethylamine (Et₃N), pyridine, diisopropylethylamine, diisopropylamine, and acetonitrile were distilled from CaH₂ at atmospheric pressure. Silica gel (0.040–0.063) by Merck was used for flash chromatography. Microanalyses were performed by Atlantic Microlabs, Atlanta, GA. Other general experimental details have been described.¹²

(2*S*)-1-[*N*-Benzyl-*N*-[3-(*tert*-butoxycarbonyl)aminopropyl]amino]-4-(*tert*-butoxycarbonyl)aminobutan-2-ol (10). (*R*)-Epichlorohydrin (**5**, 310 μL, 4.0 mmol) and [*N*-(*tert*-butoxycarbonyl)-3-aminopropyl]benzylamine (**6**, 1.00 g, 3.78 mmol) were mixed and maintained at room temperature for 24 h. Tetra-*n*-butylammonium cyanide (1.22 g, 4.54 mmol) and toluene (19 mL) were then added, and the resulting solution was heated at 70 °C for 5 h. The solution was allowed to cool to room temperature and then partitioned between EtOAc (100 mL) and brine (40 mL). The organic phase was dried (MgSO₄), filtered and concentrated. The resulting crude oil was purified by flash chromatography (7:3 hexanes–EtOAc → 6:4 hexanes–EtOAc) to give 1.05 g (80%) of nitrile **7** as a clear oil: ¹H NMR (500 MHz, CDCl₃) δ 7.23–7.32 (m, 5 H), 4.81 (s, 1H), 3.84–3.89 (m, 1 H), 3.69 (d, *J* = 13.4 Hz, 1 H), 3.49 (d, *J* = 13.4 Hz, 1 H), 3.09–3.11 (m, 2 H) 2.38–2.60 (m, 7 H), 1.61–1.67 (m, 2 H), 1.41 (s, 9 H); ¹³C NMR (125 MHz, CDCl₃)

155.9, 138.0, 128.8, 128.4, 127.3, 117.3, 79.0, 64.0, 59.0, 58.9, 51.7, 38.3, 28.3, 27.1, 23.2 ppm; IR (film) 3371, 2252, 1694 cm⁻¹; [α]_D²⁵ +41.7, [α]_D²⁵ +43.4, [α]_D²⁵ +49.5, [α]_D²⁵ +85.0, [α]_D²⁵ +102 (c 1.6, CHCl₃). Anal. Calcd for C₁₉H₂₉O₃N₃: C, 65.68; H, 8.41; N, 12.09. Found: C, 65.57; H, 8.49; N, 12.03.

Following the general procedure of Ward,^{8b} **7** was treated with (*S*)-α-methoxy-α-(trifluoromethyl)phenylacetic acid chloride [(*S*)-MTPACl] to give the corresponding (*R*)-MTPA ester. Analysis by ¹⁹F NMR showed a ratio of approximately 74:1 of 7-(*R*)-MTPA and *ent*-7-(*R*)-MTPA¹³ corresponding to 97% ee. Diagnostic ¹⁹F NMR (400 MHz, CDCl₃) signals:¹⁴ 7-(*R*)-MTPA, δ -71.4; *ent*-7-(*R*)-MTPA, δ -71.6.

(2*S*)-1-[*N*-Benzyl-*N*-[3-(*tert*-butoxycarbonyl)aminopropyl]amino]-4-(*tert*-butoxycarbonyl)aminobutan-2-ol (10). A solution of **7** (400 mg, 1.15 mmol) and Et₂O (5 mL) was added dropwise to a stirring suspension of LiAlH₄ (180 mg, 4.74 mmol) and Et₂O (15 mL) at 0 °C. After 1 h, H₂O (180 μL), 3 N NaOH (180 μL), and H₂O (540 μL) were added sequentially to the reaction mixture. The resulting mixture was filtered through a pad of Celite and concentrated, and the resulting colorless oil was used without further purification.

A mixture of this crude primary amine, BOC₂O (310 mg, 1.4 mmol), CH₂Cl₂ (5 mL), and saturated aqueous NaHCO₃ (5 mL) was stirred at room temperature for 1 h. The reaction mixture was then poured into Et₂O (20 mL), and the layers were separated. The organic layer was washed with brine (5 mL), dried (MgSO₄), and concentrated. The residual oil was purified by flash chromatography (1:1 hexanes–EtOAc) to provide 470 mg (91%) of **10** as a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 7.20–7.29 (m, 5 H), 5.15 (s, 1 H), 4.88 (s, 1 H), 3.66–3.74 (m, 2 H), 3.58 (s, 1 H), 3.30–3.39 (m, 2 H), 3.06–3.15 (m, 3 H), 2.53–2.59 (m, 1 H), 2.35–2.42 (m, 3 H), 1.52–1.62 (m, 4 H), 1.34 (s, 18 H); ¹³C NMR (75 MHz, CDCl₃) 156.2, 155.9, 138.4, 128.9, 128.3, 127.2, 78.8, 66.2, 60.2, 58.8, 51.4, 38.5, 37.9, 34.4, 28.3, 27.1 ppm; IR (film) 3354, 1694 cm⁻¹; [α]_D²⁵ +42.3, [α]_D²⁵ +44.3, [α]_D²⁵ +49.7, [α]_D²⁵ +88.2 (c 3.5, CHCl₃). Anal. Calcd for C₂₄H₄₁O₅N₃: C, 63.83; H, 9.15; N, 9.30. Found: C, 63.87; H, 9.22; N, 9.22.

(2*S*)-1-[*N*-[3-(*tert*-butoxycarbonyl)aminopropyl]amino]-4-(*tert*-butoxycarbonyl)aminobutan-2-ol (2). A mixture of **10** (120 mg, 0.27 mmol), 10% Pd/C (20 mg), and MeOH (5 mL) was maintained at room temperature under 1 atm of H₂ for 5 h. The mixture was then filtered through a plug of Celite, the plug was washed with MeOH (20 mL), and the eluent was concentrated to yield 96 mg (99%) of **2** as a colorless oil that was sufficiently pure to be employed directly in our syntheses of **3** and **4**.^{3,4} For characterization purposes, a sample of **2** was purified by flash chromatography (9:1 CHCl₃–MeOH) to give a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 5.16 (s, 1 H), 5.06 (s, 1 H), 3.69–3.71 (m, 1 H), 3.34–3.36 (m, 1 H), 3.10–3.20 (m, 3 H), 2.96 (br s, 2 H), 2.59–2.69 (m, 3 H), 2.47–2.51 (m, 1 H), 1.53–1.64 (m, 3 H), 1.39–1.50 (m, 19 H); ¹³C NMR (125 MHz, CDCl₃) 156.6, 156.2, 79.1, 79.0, 67.4, 55.2, 46.8, 38.5, 37.4, 35.0, 29.9, 28.3 ppm; IR (film) 3344, 1694 cm⁻¹; HRMS (FAB) (MH) *m/z* 362.2653 (362.2655 calcd for C₁₇H₃₆N₃O₅); [α]_D²⁵ +9.8, [α]_D²⁵ +10.2, [α]_D²⁵ +11.4, [α]_D²⁵ +18.2, [α]_D²⁵ +21.1 (c 2.0, CHCl₃).

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